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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.004 Å R factor = 0.042 wR factor = 0.109 Data-to-parameter ratio = 9.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

11β-Hydroxypregna-1,4,16-triene-3,20-dione

The steroidal title compound, $C_{21}H_{26}O_3$, possesses normal geometrical parameters. Ring *A* has a 1 α -sofa conformation, and rings *B* and *C* are in nearly perfect chair conformations. An O-H···O hydrogen bond helps to establish the crystal packing.

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Comment

The title compound (I), which can be made from 16dehydroprogesterone in a multi-step process (Herzog *et al.*, 1959), is an important steroid derivative. It has been used in the syntheses of some corticosteroids (Conrow, 1999). Its ¹H and ¹³C NMR spectra have previously been used to establish the atomic connectivity (Kirk *et al.*, 1990). Here, the singlecrystal structure of (I) (Fig. 1) is described. It confirms the spectroscopic structure.



A typical steroid conformation is observed, with three sixmembered rings and one five-membered ring denoted as rings A (C1–C6), B (C1, C6–C10), C (C9–C14) and D (C13–C17), respectively. Ring A has a 1 α -sofa conformation, and rings Band C are in nearly ideal chair conformations. Ring D is an envelope with C14 in the flap position. The configurations of the chiral C atoms in (I) (C1 R; C9 S; C10 S; C11 R; C13 S; C14 S) are consistent with those of the equivalent atoms in the starting material of known absolute configuration.

An $O-H\cdots O$ hydrogen bond between the 11-OH group of one molecule and the 3-carbonyl group of an adjacent molecule (Table 1 and Fig. 2) helps to establish the crystal packing in (I).

Experimental

Chromium (1.5 g) and hydrochloric acid (40 ml, 8.0%) were reacted under nitrogen at 348 K to generate a divalent chromium solution. A stirred solution of 16α , 17α -epoxypregna- 11α -hydroxy-1,4-diene-3,20dione (5.0 g, 14.7 mmol), provided by Tianjin Tianyao Pharmaceutical Co. Ltd, and the chromium(II) solution were then mixed and heated to 353 K for 30 min. The resulting light-yellow solid was collected by filtration and dried under vacuum at 333 K. The powder

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Figure 1

The molecular structure of (I), showing 30% displacement ellipsoids (arbitrary spheres for the H atoms).

was dissolved in ethanol and crystals of (I) grew on slow evaporation of the solvent (m.p. 547 K).

Z = 4

 $D_r = 1.198 \text{ Mg m}^{-3}$

Mo Ka radiation

Blade, light yellow

 $0.37 \times 0.13 \times 0.04 \text{ mm}$

14741 measured reflections

1959 independent reflections

1544 reflections with $I > 2\sigma(I)$

 $\mu = 0.08 \text{ mm}^-$ T = 293 (2) K

 $R_{\rm int} = 0.070$

 $\theta_{\rm max} = 25.5^\circ$

Crystal data

 $\begin{array}{l} C_{21}H_{26}O_3 \\ M_r = 326.42 \\ \text{Orthorhombic, } P2_12_12_1 \\ a = 6.2265 \ (12) \ \text{\AA} \\ b = 14.862 \ (3) \ \text{\AA} \\ c = 19.561 \ (4) \ \text{\AA} \\ V = 1810.1 \ (6) \ \text{\AA}^3 \end{array}$

Data collection

Rigaku R-AXIS RAPID IP diffractometer ω scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{\min} = 0.972, T_{\max} = 0.998$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0613P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.042$	+ 0.0192P]
$wR(F^2) = 0.109$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.10	$(\Delta/\sigma)_{\rm max} = 0.014$
1959 reflections	$\Delta \rho_{\rm max} = 0.16 \ {\rm e} \ {\rm \AA}^{-3}$
218 parameters	$\Delta \rho_{\rm min} = -0.13 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	Extinction correction: SHELXL97
	Extinction coefficient: 0.024 (5)

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
$O2-H2B\cdots O1^i$	0.82	1.93	2.733 (3)	166
	1	. 1		

Symmetry code: (i) -x + 3, $y - \frac{1}{2}$, $-z + \frac{1}{2}$.

In the absence of significant anomalous scattering effects, Friedel pairs were merged prior to refinement. The absolute configuration of (I) was assigned based on that of the well defined starting material. H atoms were placed in calculated positions (O-H = 0.82 Å, C-H =



Figure 2

The molecular packing of compound (I), viewed along the a axis. Hydrogen bonds and close contacts are shown as dashed lines.

0.93–0.98 Å) and refined as riding, with $U_{iso}(H) = 1.2U_{eq}(\text{carrier})$ or $1.5U_{eq}(\text{methyl carrier})$.

Data collection: *RAPID-AUTO* (Rigaku, 1999); cell refinement: *RAPID-AUTO*; data reduction: *RAPID-AUTO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997*a*); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXTL*.

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References

Conrow, R. E. (1999). J. Org. Chem. 64, 1042-1044.

Herzog, H. L. (1959). US Patent 2 874 172.

Kirk, D. N., Toms, H. C., Douglas, C., White, K. A., Smith, K. E., Latif, S. & Hubbard, R. W. P. (1990). J. Chem. Soc. Perkin Trans. 2, pp. 1567–1594.

Rigaku (1999). RAPID-AUTO. Rigaku Corporation, Tokyo, Japan.

Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

Sheldrick, G. M. (1997a). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Sheldrick, G. M. (1997b). SHELXTL. Bruker AXS Inc., Madison. Wisconsin, USA.